

## NOTES

**New Ravidomycin Analogues, FE35A and FE35B, Apoptosis Inducers Produced by *Streptomyces rochei***

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Apoptosis is a physiological process by which cells undergo controlled cell death accompanied by nuclear condensation and fragmentation without loss of membrane integrity<sup>1,2</sup>. This phenomenon is generally observed during the development of lymphocytes and neutrophils<sup>3</sup> in a hematopoietic system. In addition, neoplastic cells are often subjected to apoptosis when treated with anti-cancer drugs<sup>4,5</sup>. Since interleukin-1 $\beta$ -converting enzyme (ICE) proteases, particularly caspase-3, are activated during apoptosis, their overexpression

inducing cell apoptosis is considered to play an important role in the initiation of apoptosis in variety of cell systems even though their molecular mechanisms are not well understood.

During the course of our screening program for new antitumor antibiotics capable of inducing apoptosis in tumor cells, we isolated 6-hydroxytetrangulol<sup>6</sup> as a caspase-3 inducer. Further investigation has resulted in the isolation of two novel substances belonging to the benzo[d]naphtho[1,2-b]pyran-6-one group. We report herein the fermentation, isolation and structure determination of these two products designated FE35A (**1**) and FE35B (**2**).

The producing organism of **1** and **2** identified as *Streptomyces rochei* 3218-GM2, which was isolated from a soil sample collected in Iou-island, Tokyo, was cultivated at 27°C for 5 days in two 50-liter jar fermenters containing 30 liters of a medium consisting of soluble starch 2.5%, soybean meal 1.5%, dry yeast 0.2% and calcium carbonate 0.4% (pH 7.0).

The mycelial cake obtained from the cultured broth (60 liters) was extracted with acetone. After concentrated *in vacuo*, the extract was partitioned between EtOAc and H<sub>2</sub>O at pH 7.0 and the organic layer was dried over Na<sub>2</sub>SO<sub>4</sub>. The filtrate of the cultured broth was adsorbed on a Diaion HP-20 column. The column was washed with H<sub>2</sub>O and 50% MeOH, and active materials

Table 1. Physico-chemical properties of FE35A (**1**) and FE35B (**2**).

	FE35A ( <b>1</b> )	FE35B ( <b>2</b> )
Appearance	yellow powder	yellow powder
$[\alpha]_D^{23}$	-41° (c 0.05, MeOH)	-98° (c 0.05, CHCl <sub>3</sub> )
MP (dec, °C)	228~231	239~242
Formula	C <sub>30</sub> H <sub>31</sub> NO <sub>9</sub>	C <sub>31</sub> H <sub>31</sub> NO <sub>10</sub>
HRFAB-MS ( <i>m/z</i> )	550.2099 (M+H) <sup>+</sup>	578.2054 (M+H) <sup>+</sup>
calcd.	550.2078	578.2026
UV $\lambda_{max}$ ( $\epsilon$ )	245 (35100), 277 (sh),	245 (35000), 277 (sh),
in MeOH	285 (32900), 321 (13100), 334 (11200), 350 (9400), 393 (12900)	285 (32500), 321(13000), 334 (11000), 350 (9200), 393 (12900)
IR $\nu_{max}$ (KBr) cm <sup>-1</sup>	3400, 1710, 1620, 1605, 1590	3400, 1720, 1620, 1605, 1590
Solubility		
soluble in	MeOH, DMSO	CHCl <sub>3</sub> , MeOH, DMSO
insoluble in	H <sub>2</sub> O, hexane	H <sub>2</sub> O, hexane

Table 2.  $^{13}\text{C}$  (125 MHz) and  $^1\text{H}$  (500 MHz) chemical shift data of FE35A (**1**, in  $\text{MeOH-}d_4$ ), **B** (**2**, in  $\text{CDCl}_3$ ) and ravidomycin (in  $\text{CDCl}_3$ ).

No.	FE35A ( <b>1</b> )		FE35B ( <b>2</b> )		Ravidomycin	
	$\delta_{\text{C}}$	$\delta_{\text{H}}$	$\delta_{\text{C}}$	$\delta_{\text{H}}$	$\delta_{\text{C}}$	$\delta_{\text{H}}$
1	155.8		154.7		154.7	
2	113.3	6.99 (d, 8.3)	112.6	6.98 (d, 8.5)	112.4	7.04 (d, 8.3)
3	130.8	7.92 (d, 8.3)	129.5	7.78 (d, 8.5)	129.5	7.98 (d, 8.3)
4	126.7		124.5		125.2	
4a	124.5		125.5		125.2	
4b	143.8		142.5		142.9	
6	161.9		160.7		160.6	
6a	123.3		122.2		122.2	
7	120.3	7.97 (d, 1.2)	120.2	8.08 (d, 1.6)	119.7	8.02 (d, 1.6)
8	140.5		138.9		138.6	
9	115.5	7.50 (d, 1.2)	114.3	7.33 (d, 1.6)	113.8	7.25 (d, 1.6)
10	158.9		157.4		157.1	
10a	123.5		123.9		123.4	
10b	115.4		114.2		113.9	
11	103.1	8.43 (s)	102.7	8.45 (s)	102.1	8.28 (s)
12	153.4		152.2		151.8	
12a	117.0		116.5		116.3	
13	136.5	6.82 (dd, 17.5, 10.5)	135.2	6.78 (dd, 17.0, 11.0)	135.2	6.76 (dd, 17.2, 10.5)
14	117.0	6.00 (d, 17.5) 5.43 (d, 10.5)	116.6	5.91 (d, 17.0) 5.41 (d, 11.0)	116.2	5.91 (d, 17.2) 5.44 (d, 10.5)
10-OMe	56.9	4.10 (s)	56.4	4.07 (s)	56.0	4.00 (s) <sup>§</sup>
12-OMe	56.7	4.08 (s)	56.4	4.08 (s)	55.9	4.01 (s) <sup>§</sup>
1'	81.3	6.03 (d, 8.8)	79.9	5.85 (d, 9.3)	80.5	5.85 (d, 9.4)
2'	67.5	4.46 (dd, 9.7, 8.8)	71.2	4.08 (dd, 9.4, 9.3)	65.4	4.44 (dd, 10.2, 9.4)
3'	62.8	4.71 (dd, 9.7, 2.2)	57.1	4.48 (m)	69.6	3.08 (dd, 10.2, 3.0)
4'	74.2	3.92 (d, 2.2)	74.2	5.26 (d, 2.2)	69.4	5.56 (d, 3.0)
5'	77.0	4.40 (q, 6.4)	73.8	4.51 (q, 6.4)	75.0	4.49 (q, 6.4)
6'	17.0	1.20 (d, 6.4)	16.9	1.09 (d, 6.4)	16.8	1.05 (d, 6.4)
3'-NMe	34.0	3.24 (s)			40.8	2.51 (s)
3'-NCOMe	22.4	2.17 (s)	23.3	2.01 (s)		
3'-NCOMe	174.6		172.0			
4'-COMe			20.9	2.17 (s)	21.6	2.11 (s)
4'-COMe			172.5		170.8	
1-OH			9.78 (s)			9.77 (s)
3'-NH			6.05 (brd, 7.0)			

Chemical shifts are expressed in ppm relative to TMS as an internal standard.

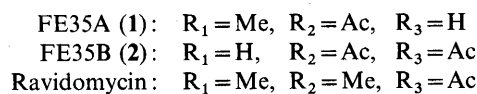
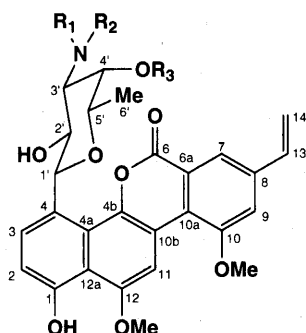
§: Assignments may be interchangeable.

were eluted with MeOH. The EtOAc extract and the column eluate were combined and concentrated to dryness, and the residue was subjected to silica gel column chromatography developed with  $\text{CHCl}_3$  - MeOH (50:1) to give active fractions. They were combined and applied to a Toyopearl HW-40 column packed with MeOH to yield a mixture of **1** and **2**. This material was subjected to preparative HPLC using ODS with 45% acetonitrile and 50 mM disodium citrate (pH 4.0) to give fractions containing pure **1** or **2**. After concentration *in vacuo*, the aqueous residue containing **1** was extracted

with EtOAc three times. The combined extract was concentrated to give a pure sample of **1** (2.0 mg) as a yellow amorphous powder. In a similar manner, **2** was obtained at the yield of 2.3 mg.

The physico-chemical properties, and the  $^1\text{H}$  and  $^{13}\text{C}$  NMR data of **1** and **2** are summarized in Table 1 and 2, respectively. **1** was soluble in MeOH and DMSO, slightly soluble in  $\text{CHCl}_3$  and EtOAc, but insoluble in water and *n*-hexane. **2** was soluble in  $\text{CHCl}_3$ , MeOH and DMSO but insoluble in water and *n*-hexane. The molecular formulae of **1** and **2** were established as

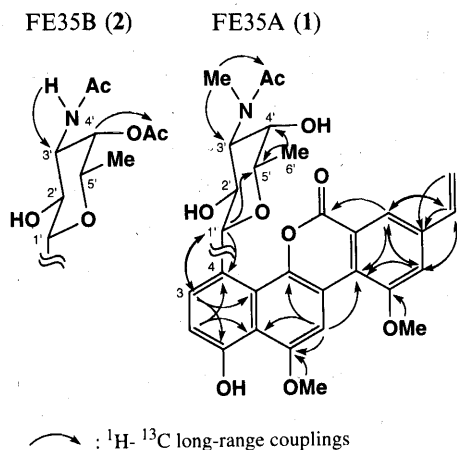
Fig. 1. Structures of FE35A (1), FE35B (2) and ravidomycin.



C<sub>30</sub>H<sub>31</sub>NO<sub>9</sub> and C<sub>31</sub>H<sub>31</sub>NO<sub>10</sub>, respectively, from HRFAB-MS and NMR spectral analyses. IR absorptions at 3400 cm<sup>-1</sup> and 1710 (1720 in **2**) cm<sup>-1</sup> suggested the presence of hydroxyl and ester functions in **1** and **2**. Since the UV and visible spectra of **1** and **2** resembled those of ravidomycins<sup>7,8</sup>, chrysomycins<sup>9</sup> and gilvocarcins<sup>10</sup>, the chromophore of **1** and **2** was revealed to be benzo[d]naphtho[1,2-b]pyran-6-one as shown in Fig. 1. Long-range couplings from aromatic protons observed in the HMBC spectra also supported the chromophore structure of **1** and **2** (Fig. 2).

The structure of **1** was elucidated by NMR analysis as follows. In the HMBC spectrum of **1**, a vinyl proton 13-H (6.82 ppm) coupled to *exo* methylene protons 14-H (6.00 ppm; 17.5 Hz, 5.43 ppm; 10.5 Hz), was long-range coupled to aromatic carbons C-7 (120.3 ppm), C-8 (140.5 ppm) and C-9 (115.5 ppm). Thus, the connectivity between the aglycone and the vinyl residue was proved as shown in Fig. 2. Likewise, two methoxy groups were linked to C-10 (158.9 ppm) and C-12 (153.4 ppm). These partial structures together with similar UV and visible spectral features between **1** and ravidomycin suggested the presence of the same chromophore in both compounds. The <sup>1</sup>H and <sup>13</sup>C NMR data summarized in Table 2 corroborated this conclusion. The structural difference between ravidomycin and **1** was ascribed to the sugar moiety. NMR spectral analysis of the sugar group of **1** revealed a C-glycosylated 6-deoxyhexopyranose. The sequence from 1'-H (6.03 ppm) to 4'-H (3.92 ppm) through 2'-H (4.46 ppm) and 3'-H (4.71 ppm) was established by <sup>1</sup>H-<sup>1</sup>H COSY spectral analysis. The connectivity between C-4' (74.2 ppm), C-5' (77.0 ppm)

Fig. 2. NMR analyses of **1** and **2**.



and C-6' (17.0 ppm) was determined by long-range couplings from 6'-H methyl protons (1.20 ppm) to C-4' and C-5'. The tetrahydropyran moiety was elucidated by a long-range coupling from 1'-H to C-5'. Long-range couplings from a methylamino proton 3'-NMe (3.42 ppm) to C-3' and an acetyl carbon 3'-NCO (174.6 ppm) were attributed to the existence of an *N,N*-acetyl methylamino group at C-3' position. Large coupling constants of 2'-H ( $J=8.8, 9.7$  Hz) revealed the relative configuration of 1'-, 2'- and 3'- protons on the tetrahydropyran to be axial-axial. A small coupling constant ( $J=2.2$  Hz) between 3'-H and 4'-H (3.92 ppm) showed the relative configuration to be axial-equatorial. On the other hand, negligible coupling between H-4' and H-5' (4.40 ppm) indicated that 4'-H and 5'-H have an equatorial and an axial configuration, respectively<sup>7,8</sup>. The <sup>13</sup>C chemical shift of C-6' (17.0 ppm) similar to that of ravidomycin (16.8 ppm) also supported the relative configuration of 5'-H to be axial. Finally, the C-glycosidic moiety was combined to the C-4 position by long-range couplings from 1'-H to C-3 (130.8 ppm) and C-4 (126.7 ppm), and 3-H to C-1'. Thus, the remaining hydroxyl residue was substituted to the C-1 (155.8 ppm) position as seen in ravidomycin. Other long-range couplings observed in the HMBC spectrum together with UV and visible spectra ascribed the structure of **1** as shown in Fig. 1.

The analyses of the HMBC spectral data of **2** confirmed the existence of the same chromophore as those of **1** and ravidomycins. In addition, a tetrahydropyran moiety was linked to the chromophore at C-4 position through a C-glycosidic linkage. The differences between **1** and **2** are ascribed to the substituents at C-3' and C-4' (69.4 ppm)

on the tetrahydropyran moiety. The methylamino signal in **1** was replaced by an amino proton (6.05 ppm) which was long-range coupled to C-3' (69.6 ppm) and an amino acyl carbon (172.0 ppm) in **2**. Low field shift of H-4' (5.26 ppm) revealed the sugar moiety acetylated at the C-4' position of **2**. Other NMR data including long-range couplings from a phenolic hydroxyl proton 1-OH (9.78 ppm) to aromatic carbons C-1 (154.7 ppm), C-2 (112.6 ppm) and C-12a (116.5 ppm) established the structure of **2** as shown in Fig. 2.

**1** and **2** showed cytotoxicity against U937 leukemia cells with the IC<sub>50</sub> values of 3.7 and 4.6 nM, respectively. In order to investigate the relationship between the cytotoxic activities of these compounds and the ability to induce caspase-3, we employed a system using DEVD-AMC (*N*-acetyl-Asp-Glu-Val-Asp-7-amino-4-methylcoumarin) as the caspase-3 substrate<sup>6)</sup>. Caspase-3 induced by apoptotic agents cleaves DEVD-AMC to AMC in this assay system enabling estimation of the caspase-3 activity is estimated by measuring the UV and visible absorption of AMC. In this evaluation system, **1**, **2** and ravidomycin induced a caspase-3 like activity in U937 leukemia cells at concentrations higher than 200 nM for 3 hours after administrations.

Ravidomycin and other coumarin-related antitumor agents have been reported as potent inhibitors of human topoisomerase II<sup>11)</sup>. Dimethylepipodophyllotoxin VP-16 and adriamycin are known as clinical important antitumor drugs that also inhibit eukaryotic topoisomerase II, which were reported to induce the CPP32 like proteases in tumor cells<sup>12)</sup>. Further studies on detailed biological activities of **1** and **2** are now underway.

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